

ISSN 2518-1491 (Online),  
ISSN 2224-5286 (Print)

ҚАЗАҚСТАН РЕСПУБЛИКАСЫ  
ҰЛТТЫҚ ҒЫЛЫМ АКАДЕМИЯСЫНЫҢ  
Д.В. Сокольский атындағы  
«Жанармай, катализ және электрохимия институты» АҚ

# Х А Б А Р Л А Р Ы

---

---

## ИЗВЕСТИЯ

НАЦИОНАЛЬНОЙ АКАДЕМИИ НАУК  
РЕСПУБЛИКИ КАЗАХСТАН  
АО «Институт топлива, катализа и  
электрохимии им. Д.В. Сокольского»

## N E W S

OF THE ACADEMY OF SCIENCES  
OF THE REPUBLIC OF KAZAKHSTAN  
JSC «D.V. Sokolsky institute of fuel, catalysis  
and electrochemistry»

**SERIES**  
**CHEMISTRY AND TECHNOLOGY**

**3 (447)**

**MAY – JUNE 2021**

PUBLISHED SINCE JANUARY 1947

PUBLISHED 6 TIMES A YEAR

ALMATY, NAS RK

---

---

*NAS RK is pleased to announce that News of NAS RK. Series of chemistry and technologies scientific journal has been accepted for indexing in the Emerging Sources Citation Index, a new edition of Web of Science. Content in this index is under consideration by Clarivate Analytics to be accepted in the Science Citation Index Expanded, the Social Sciences Citation Index, and the Arts & Humanities Citation Index. The quality and depth of content Web of Science offers to researchers, authors, publishers, and institutions sets it apart from other research databases. The inclusion of News of NAS RK. Series of chemistry and technologies in the Emerging Sources Citation Index demonstrates our dedication to providing the most relevant and influential content of chemical sciences to our community.*

*Қазақстан Республикасы Ұлттық ғылым академиясы «ҚР ҰҒА Хабарлары. Химия және технология сериясы» ғылыми журналының Web of Science-тің жаңаланған нұсқасы Emerging Sources Citation Index-те индекстелуге қабылданғанын хабарлайды. Бұл индекстелу барысында Clarivate Analytics компаниясы журналды одан әрі the Science Citation Index Expanded, the Social Sciences Citation Index және the Arts & Humanities Citation Index-ке қабылдау мәселесін қарастыруда. Web of Science зерттеушілер, авторлар, баспашылар мен мекемелерге контент тереңдігі мен сапасын ұсынады. ҚР ҰҒА Хабарлары. Химия және технология сериясы Emerging Sources Citation Index-ке енуі біздің қоғамдастық үшін ең өзекті және беделді химиялық ғылымдар бойынша контентке адалдығымызды білдіреді.*

*НАН РК сообщает, что научный журнал «Известия НАН РК. Серия химии и технологий» был принят для индексирования в Emerging Sources Citation Index, обновленной версии Web of Science. Содержание в этом индексировании находится в стадии рассмотрения компанией Clarivate Analytics для дальнейшего принятия журнала в the Science Citation Index Expanded, the Social Sciences Citation Index и the Arts & Humanities Citation Index. Web of Science предлагает качество в глубину контента для исследователей, авторов, издателей и учреждений. Включение Известия НАН РК в Emerging Sources Citation Index демонстрирует нашу приверженность к наиболее актуальному и влиятельному контенту по химическим наукам для нашего сообщества.*

### **Бас редактор:**

**ЖҰРЫНОВ Мұрат Жұрынұлы**, химия ғылымдарының докторы, профессор, ҚР ҰҒА академигі, Қазақстан Республикасы Ұлттық ғылым академиясының президенті, АҚ «Д.В. Сокольский атындағы отын, катализ және электрохимия институтының» бас директоры (Алматы, Қазақстан) Н = 4

### **Редакция алқасы:**

**ӘДЕКЕНОВ Серғазы Мыңжасарұлы** (бас редактордың орынбасары), химия ғылымдарының докторы, профессор, ҚР ҰҒА академигі, «Фитохимия» Халықаралық ғылыми-өндірістік холдингінің директоры (Қарағанды, Қазақстан) Н = 11

**АГАБЕКОВ Владимир Енокович** (бас редактордың орынбасары), химия ғылымдарының докторы, профессор, Беларусь ҰҒА академигі, Жаңа материалдар химиясы институтының құрметті директоры (Минск, Беларусь) Н = 13

**СТРНАД Мирослав**, профессор, Чехия ғылым академиясының Эксперименттік ботаника институтының зертхана меңгерушісі (Оломоуц, Чехия) Н = 66

**БҮРКІТБАЕВ Мұхамбетқали**, химия ғылымдарының докторы, профессор, ҚР ҰҒА академигі, әл-Фараби атындағы ҚазҰУ-дың бірінші проректоры (Алматы, Қазақстан) Н = 11

**ХОХМАНН Джудит**, Сегед университетінің Фармацевтика факультетінің Фармакогнозия кафедрасының меңгерушісі, Жаратылыстану ғылымдарының пәнаралық орталығының директоры (Сегед, Венгрия) Н = 38

**РОСС Самир, PhD докторы**, Миссисипи университетінің Өсімдік өнімдерін ғылыми зерттеу ұлттық орталығы, Фармация мектебінің профессоры (Оксфорд, АҚШ) Н = 35

**ХУТОРЯНСКИЙ Виталий**, философия докторы (PhD, фармацевт), Реддинг университетінің профессоры (Реддинг, Англия) Н = 40

**ТЕЛТАЕВ Бағдат Бұрханбайұлы**, техника ғылымдарының докторы, профессор, ҚР ҰҒА корреспондент-мүшесі, Қазақстан Республикасы Индустрия және инфрақұрылымдық даму министрлігі (Алматы, Қазақстан) Н = 13

**ФАРУК Асана Дар**, Хамдар аль-Маджида Шығыс медицина колледжінің профессоры, Хамдар университетінің Шығыс медицина факультеті (Карачи, Пәкістан) Н = 21

**ФАЗЫЛОВ Серік Драхметұлы**, химия ғылымдарының докторы, профессор, ҚР ҰҒА академигі, Органикалық синтез және көмір химиясы институты директорының ғылыми жұмыстар жөніндегі орынбасары (Қарағанды, Қазақстан) Н = 6

**ЖОРОБЕКОВА Шарипа Жоробекқызы**, химия ғылымдарының докторы, профессор, Қырғызстан ҰҒА академигі, ҚР ҰҒА Химия және химиялық технология институты (Бішкек, Қырғызстан) Н = 4

**ХАЛИКОВ Джурабай Халикович**, химия ғылымдарының докторы, профессор, Тәжікстан ҒА академигі, В.И. Никитин атындағы Химия институты (Душанбе, Тәжікстан) Н = 6

**ФАРЗАЛИЕВ Вагиф Меджидоглы**, химия ғылымдарының докторы, профессор, ҰҒА академигі (Баку, Әзірбайжан) Н = 13

**ГАРЕЛИК Хемда**, философия докторы (PhD, химия), Халықаралық таза және қолданбалы химия одағының Химия және қоршаған орта бөлімінің президенті (Лондон, Англия) Н = 15

### **«ҚР ҰҒА Хабарлары. Химия және технология сериясы»**

**ISSN 2518-1491 (Online),**

**ISSN 2224-5286 (Print)**

Меншіктенуші: «Қазақстан Республикасының Ұлттық ғылым академиясы» РҚБ (Алматы қ.).

Қазақстан Республикасының Ақпарат және қоғамдық даму министрлігінің Ақпарат комитетінде 29.07.2020 ж. берілген № **KZ66VPY00025419** мерзімдік басылым тіркеуіне қойылу туралы куәлік.

Тақырыптық бағыты: *химия және жаңа материалдар технологиясы саласындағы басым ғылыми зерттеулерді жариялау.*

Мерзімділігі: жылына 6 рет.

Тиражы: 300

Редакцияның мекен-жайы: 050010, Алматы қ., Шевченко көш., 28, 219 бөл., тел.: 272-13-19, 272-13-18  
<http://chemistry-technology.kz/index.php/en/archiv>

© Қазақстан Республикасының Ұлттық ғылым академиясы, 2021

---

Редакцияның мекенжайы: 050100, Алматы қ., Қонаев к-сі, 142, «Д. В. Сокольский атындағы отын, катализ және электрохимия институты» АҚ, каб. 310, тел. 291-62-80, факс 291-57-22, e-mail:orgcat@nursat.kz

Типографияның мекен-жайы: «Аруна» ЖК, Алматы қ., Муратбаева көш., 75.

### Главный редактор:

**ЖУРИНОВ Мурат Журинович**, доктор химических наук, профессор, академик НАН РК, президент Национальной академии наук Республики Казахстан, генеральный директор АО «Институт топлива, катализа и электрохимии им. Д.В. Сокольского» (Алматы, Казахстан) Н = 4

### Редакционная коллегия:

**АДЕКЕНОВ Сергазы Мынжасарович** (заместитель главного редактора), доктор химических наук, профессор, академик НАН РК, директор Международного научно-производственного холдинга «Фитохимия» (Караганда, Казахстан) Н = 11

**АГАБЕКОВ Владимир Енокович** (заместитель главного редактора), доктор химических наук, профессор, академик НАН Беларуси, почетный директор Института химии новых материалов (Минск, Беларусь) Н = 13

**СТРНАД Мирослав, профессор**, заведующий лабораторией института Экспериментальной ботаники Чешской академии наук (Оломоуц, Чехия) Н = 66

**БУРКИТБАЕВ Мухамбеткали**, доктор химических наук, профессор, академик НАН РК, Первый проректор КазНУ имени аль-Фараби (Алматы, Казахстан) Н = 11

**ХОХМАНН Джудит**, заведующий кафедрой Фармакогнозии Фармацевтического факультета Университета Сегеда, директор Междисциплинарного центра естественных наук (Сегед, Венгрия) Н = 38

**РОСС Самир**, доктор PhD, профессор Школы Фармации национального центра научных исследований растительных продуктов Университета Миссисипи (Оксфорд, США) Н = 35

**ХУТОРЯНСКИЙ Виталий**, доктор философии (Ph.D, фармацевт), профессор Университета Рединга (Рединг, Англия) Н = 40

**ТЕЛЬТАЕВ Багдат Бурханбайулы**, доктор технических наук, профессор, член-корреспондент НАН РК, Министерство Индустрии и инфраструктурного развития Республики Казахстан (Алматы, Казахстан) Н = 13

**ФАРУК Асана Дар**, профессор колледжа Восточной медицины Хамдарда аль-Маджида, факультет Восточной медицины университета Хамдарда (Карачи, Пакистан) Н = 21

**ФАЗЫЛОВ Серик Драхметович**, доктор химических наук, профессор, академик НАН РК, заместитель директора по научной работе Института органического синтеза и углехимии (Караганда, Казахстан) Н = 6

**ЖОРОБЕКОВА Шарипа Жоробековна**, доктор химических наук, профессор, академик НАН Кыргызстана, Институт химии и химической технологии НАН КР (Бишкек, Кыргызстан) Н = 4

**ХАЛИКОВ Джурабай Халикович**, доктор химических наук, профессор, академик АН Таджикистана, Институт химии имени В.И. Никитина АН РТ (Душанбе, Таджикистан) Н = 6

**ФАРЗАЛИЕВ Вагиф Меджид оглы**, доктор химических наук, профессор, академик НАНА (Баку, Азербайджан) Н = 13

**ГАРЕЛИК Хемда**, доктор философии (Ph.D, химия), президент Отдела химии и окружающей среды Международного союза чистой и прикладной химии (Лондон, Англия) Н = 15

«Известия НАН РК. Серия химии и технологий».

ISSN 2518-1491 (Online),

ISSN 2224-5286 (Print)

Собственник: Республиканское общественное объединение «Национальная академия наук Республики Казахстан» (г. Алматы).

Свидетельство о постановке на учет периодического печатного издания в Комитете информации Министерства информации и общественного развития Республики Казахстан № KZ66VPY00025419, выданное 29.07.2020 г.

Тематическая направленность: *публикация приоритетных научных исследований в области химии и технологий новых материалов.*

Периодичность: 6 раз в год.

Тираж: 300

Адрес редакции: 050010, г. Алматы, ул. Шевченко, 28, оф. 219, тел.: 272-13-19, 272-13-18

<http://chemistry-technology.kz/index.php/en/arithiv>

© Национальная академия наук Республики Казахстан, 2021

Адрес редакции: 050100, г. Алматы, ул. Кунаева, 142, АО «Институт топлива, катализа и электрохимии им. Д.В. Сокольского», каб. 310, тел. 291-62-80, факс 291-57-22, e-mail:orgcat@nursat.kz

Адрес типографии: ИП «Аруна», г. Алматы, ул. Муратбаева, 75.

#### Editor in chief:

**ZHURINOV Murat Zhurinovich**, doctor of chemistry, professor, academician of NAS RK, president of NAS RK, general director of JSC "Institute of fuel, catalysis and electrochemistry named after D.V. Sokolsky (Almaty, Kazakhstan) H = 4

#### Editorial board:

**ADEKENOV Sergazy Mynzhasarovich** (deputy editor-in-chief) doctor of chemical sciences, professor, academician of NAS RK, director of the international Scientific and production holding «Phytochemistry» (Karaganda, Kazakhstan) H = 11

**AGABEKOV Vladimir Enokovich** (deputy editor-in-chief), doctor of chemistry, professor, academician of NAS of Belarus, honorary director of the Institute of Chemistry of new materials (Minsk, Belarus) H = 13

**STRNAD Miroslav**, head of the laboratory of the institute of Experimental Botany of the Czech academy of sciences, professor (Olomouc, Czech Republic) H = 66

**BURKITBAYEV Mukhambetkali**, doctor of chemistry, professor, academician of NAS RK, first vice-rector of al-Farabi KazNU (Almaty, Kazakhstan) H = 11

**HOHMANN Judith**, head of the department of pharmacognosy, faculty of Pharmacy, university of Szeged, director of the interdisciplinary center for Life sciences (Szeged, Hungary) H = 38

**ROSS Samir, Ph.D.**, professor, school of Pharmacy, national center for scientific research of Herbal Products, University of Mississippi (Oxford, USA) H = 35

**KHUTORANSKY Vitaly, Ph.D.**, pharmacist, professor at the University of Reading (Reading, England) H = 40

**TELTAYEV Bagdat Burkhanbayuly**, doctor of technical sciences, professor, corresponding member of NAS RK, ministry of Industry and infrastructure development of the Republic of Kazakhstan (Almaty, Kazakhstan) H = 13

**PHARUK Asana Dar**, professor at Hamdard al-Majid college of Oriental medicine. faculty of Oriental medicine, Hamdard university (Karachi, Pakistan) H = 21

**FAZYLOV Serik Drakhmetovich**, doctor of chemistry, professor, academician of NAS RK, deputy director for institute of Organic synthesis and coal chemistry (Karaganda, Kazakhstan) H = 6

**ZHOROBEKOVA Sharipa Zhorobekovna**, doctor of chemistry, professor, academician of NAS of Kyrgyzstan, Institute of Chemistry and chemical technology of NAS KR (Bishkek, Kyrgyzstan) H = 4

**KHALIKOV Jurabay Khalikovich**, doctor of chemistry, professor, academician of the academy of sciences of Tajikistan, institute of Chemistry named after V.I. Nikitin AS RT (Tajikistan) H = 6

**FARZALIEV Vagif Medzhid ogly**, doctor of chemistry, professor, academician of NAS of Azerbaijan (Azerbaijan) H = 13

**GARELIK Hemda**, PhD in chemistry, president of the department of Chemistry and Environment of the International Union of Pure and Applied Chemistry (London, England) H = 15

#### News of the National Academy of Sciences of the Republic of Kazakhstan.

##### Series of chemistry and technology

**ISSN 2518-1491 (Online),**

**ISSN 2224-5286 (Print)**

Owner: RPA «National Academy of Sciences of the Republic of Kazakhstan» (Almaty).

The certificate of registration of a periodical printed publication in the Committee of information of the Ministry of Information and Social Development of the Republic of Kazakhstan **No. KZ66VPY00025419**, issued 29.07.2020.

Thematic scope: *publication of priority research in the field of chemistry and technology of new materials*

Periodicity: 6 times a year.

Circulation: 300

Editorial address: 28, Shevchenko str., of. 219, Almaty, 050010, tel. 272-13-19; 272-13-18,  
<http://chemistry-technology.kz/index.php/en/arhiv>

© National Academy of Sciences of the Republic of Kazakhstan, 2021

Editorial address: JSC «D.V. Sokolsky institute of fuel, catalysis and electrochemistry», 142, Kunayev str., of. 310, Almaty, 050100, tel. 291-62-80, fax 291-57-22, e-mail: [orgcat@nursat.kz](mailto:orgcat@nursat.kz)

Address of printing house: ST «Aruna», 75, Muratbayev str, Almaty.

DOI:10.1016/j.scitotenv.2020.138006 (in Eng.).

[8] Marsh H, Rodriguez-Reinoso F (2006) Activated Carbon, 1st ed.; Elsevier Science & Technology Books: Oxford, UK, ISBN: 0080444636 (in Eng.).

[9] Hagemann N, Spokas KA, Schmidt H-P, Kaegi R., Boehler M, Bucheli TD. (2018) Activated Carbon, Biochar and Charcoal: Linkages and Synergies across Pyrogenic Carbon's ABCs *Water* 10:182. DOI: 10.3390/w10020182 (in Eng.).

[10] Appazov NO, Bazarbayev BM, Assylbekkyzy T, Diyarova B M, Kanzhar SA, Magauiya S, Zhapparbergenov RU, Akyzbekov NI, Duisembekov B A. (2021) Obtaining biochar from rice husk and straw, *News of the National Academy of Sciences of the Republic of Kazakhstan*, 1:66-74.

[11] Wen Q, Li C, Cai Z, Zhang W, Gao H, Chen L, Zeng G, Shu X, Zhao Y. (2011) Study on activated carbon derived from sewage sludge for adsorption of gaseous formaldehyde, *Bioresour Technol.* 102:942–947. DOI:10.1016/j.biortech.2010.09.042 (in Eng.).

[12] Carvalho A, Mestre AS, Fernandes AC, Pinto ML. (2006) Granular activated carbons from powdered samples using clays as binders for the adsorption of organic vapours, *Microporous Mesoporous Mater* 93:226–231. DOI: 10.3390/ma13225180 (in Eng.).

[13] Lyubchyk, S., Shapovalova, O., Lygina, O., Oliveira, M. C., Appazov, N., Lyubchyk, A., Charmier, A. J., Lyubchik, S., & Pombeiro, A. J. L. (2019). Integrated Green Chemical Approach to the Medicinal Plant *Carpobrotus edulis* Processing [Article]. *Scientific Reports*, 9, 12, Article 18171. <https://doi.org/10.1038/s41598-019-53817-8> (in Eng.).

[14] Azarov VH, Zuev OV, Matvyenko VG. (2002) Method of obtaining fuel briquettes [Sposob oderzhannya palivnykh brykativ]. Preliminary Patent of the Patent Ukraine [Predvaritel'nyy patent Patenta Ukrainy]. (In Ukrainian).

[15] Azarov VH, Matvyenko VH. (2003) Method of obtaining fuel briquettes [Sposob oderzhannya palivnykh brykativ]. Preliminary Patent of the Patent Ukraine [Predvaritel'nyy patent Patenta Ukrainy]. (In Ukrainian).

[16] GOST 6217. Coal active charcoal crushed. Technical conditions [Ugol' aktivnyj drevesnyj droblenyj. Tehnicheskie uslovija]. Standartinform, Moscow, Russia, 2003. (In Russian)

## NEWS

OF THE NATIONAL ACADEMY OF SCIENCES OF THE REPUBLIC OF KAZAKHSTAN

**SERIES CHEMISTRY AND TECHNOLOGY**

ISSN 2224-5286

Volume 3, Number 447 (2021), 12 – 18

<https://doi.org/10.32014/2021.2518-1491.43>

UDC 663.223.3

**Z.A. Anarbekova\*, G.I. Baigazieva**

Almaty Technological University, Almaty, Kazakhstan.

E-mail: zaure96\_96@mail.ru

### **THE INFLUENCE OF YEAST RACES ON THE AROMA-FORMING SUBSTANCES OF TABLE WINES**

**Abstract.** Wine is a product of biochemical transformations, compounds present in grape juice, by controlled alcoholic fermentation, that is, effervescence. Grape and yeast enzymes play a key role in the processing of grapes and the preparation of wine, influencing all biotechnological processes of winemaking.

Adding liquid or dry active yeast to the wort allows better control of the fermentation process. Under the influence of these yeasts, sugar is converted mainly into alcohol or carbon dioxide, but the yeast itself during fermentation produces many molecules (higher alcohols, esters) that affect the aroma and taste of wine. These transformations take about two weeks and lead to a significant increase in temperature, which must be regulated, not allowing it to rise above 18-20°C: otherwise, some of the aromatic substances may evaporate and the fermentation process itself will stop.

The amount of yeast that determines the correct and complete fermentation depends both on the quality of the wort itself, and on the more or less prolonged access of air, the ambient temperature. The air, or rather the oxygen of the air, has a beneficial effect on fermentation as long as there are still many nutrients (sugars) in the wort; as the latter are consumed, extremely small yeast cells are formed, which persist for a long time in the form of turbidity. The rapid course of fermentation can be greatly facilitated by the periodic stirring of yeast, which, settling to the bottom, lose direct contact with nutrients — the lower layers almost do not

function. You can mix the wort mechanically or by adding healthy whole grapes to it; in this case, the wort is constantly and automatically mixed: the berries, rising up in the fermenting liquid, carry the yeast with them. In order to speed up the fermentation, the wort is sometimes ventilated, that is, air is introduced into it, by mixing.

This article shows the influence of the yeast race on the fermentation dynamics of white grape must, the composition of organic acids and aroma-forming components. The races that ensure the production of high-quality wine materials are identified.

**Key words:** Wine yeast, fermentation, enzyme activity, fermentation dynamics, organic acids, aroma-forming substances.

**Introduction.** One of the most important technological operations in the production of white table wines is the fermentation of wort, carried out by special yeast races. According to scientists [1, 2], it is the yeast races that have a great influence on the dynamics of fermentation, the formation of taste and aroma, and even the bottling resistance of wine materials. With the right selection of the race, you can produce wine material of the predicted quality and chemical composition. The modern domestic market of auxiliary materials has a large number of new yeast races. However, before recommending them for use in production, it is necessary to conduct a set of studies (tests) in order to justify their use for a certain category of wines. The main purpose of the work is to study the influence of yeast races LEVURE IOC BE THIOLS, LEVURE IOC B 2000 and LEVURE IOC B 3000 (produced in France) on the quality indicators of white table wines [5,7].

**Materials and methods.** As an object of research, the wort from the white grape variety Rkatsiteli was used. For the fermentation of grape must from white grape varieties, new races of active dry yeast were used, having the following characteristics:

LEVURE IOC B 3000 - Emphasizes the aromatic intensity and complexity of the wine with pronounced shades of yellow fruits and flowers, while contributing to the formation of a sense of volume and roundness in the taste. The strain's good fermentation abilities make it an ally of the winemaker in preventing the risks of sulfur compound odors, the so-called "reduction" tones. Ideal yeast for making elegant wines with a rounded, long-lasting taste when using yeast-lees aging technology.

LEVURE IOC B 2000 - promotes the disclosure of fruit aromatic components contained in white and pink wort. It is used for the production of fresh, aromatic wines. It allows you to obtain different aromatic profiles depending on the fermentation temperature and, thus, to harmoniously combine the fermentation tones and varietal flavors. Yeast IOC B 2000 is used in the process of vinification of white wines, for which the main aspect is the aromatic expressiveness. To an even greater extent, it is advisable to use the strain for grape varieties that are poor in the content of flavor precursors.

LEVURE IOC BE THIOLS – Preparation of yeast IOC BE THIOLS - the result of the application of an innovative method of selection using markers.

It provides the formation of thiol compounds with fruit flavors in a high concentration, and at the same time is characterized by the lack of the ability to produce  $\text{SO}_2$  and  $\text{H}_2\text{S}$ . Therefore, it is a "tool" in the hands of the winemaker who wants to reduce the amount of sulfites in the white and pink wines produced.

The volume of carbon dioxide released was determined using hydraulic locks. Mass concentration of organic acids – by capillary electrophoresis according to GOST R 52841-2007, and aroma-forming compounds-using gas-liquid chromatography by direct sample input (Crystal 2000M).

**The results of the study.** In grape must from the Rkatsiteli grape variety in the same quantities ( $2 \text{ g/dm}^3$ ) reactivated cells of yeast races LEVURE IOC BE THIOLS, LEVURE IOC B 2000, LEVURE IOC B 3000 and control - LEVURE IOC 18-2007, fermentation temperature  $18\text{-}20^\circ\text{C}$  were introduced. The initial concentration of sugars in the initial wort is  $18.5 \text{ g/100 cm}^3$ , and the titrated acids are  $8 \text{ g/dm}^3$ . Fermentation was carried out under hermetic conditions [6].

**Discussion of the results.** The analysis of the obtained data shows a difference in the dynamics of fermentation depending on the yeast race (Fig. 1).

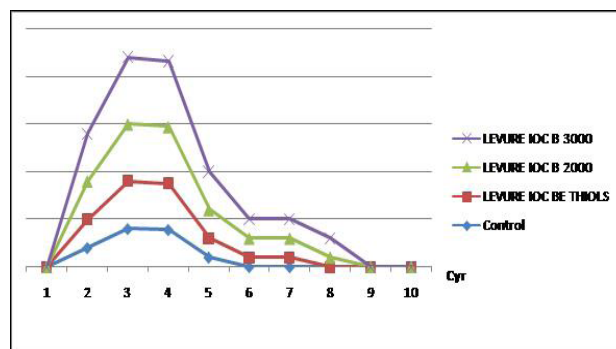


Figure 1. Dynamics of  $\text{CO}_2$  release during fermentation of wort by yeast races

The latent period was almost identical for all three experimental yeast races. In the control version, the fermentation was somewhat slower, and the exponential phase was faster. The most active sugar consumption and  $\text{CO}_2$  release were observed when using the yeast race LEVURE IOC BE THIOLS and LEVURE IOC B 2000. The least active release of  $\text{CO}_2$  was observed in the control sample. According

to the data obtained by the fermentation of sugars race LEVURE IOC THIOLS BE ended on the sixth day, other experimental races - on the ninth day, and race LEVURE IOC 18-2007 (control) for 10 days [4].

The results obtained can be explained by the difference in the activity of the enzyme systems of each of the yeast races, which is most likely a genetic feature of the strain. In this regard, it is of interest to study the chemical composition of the obtained wine materials [12]. It was found that the volume fraction of ethyl alcohol varied from 10 % vol. LEVURE IOC BE THIOLS, LEVURE IOC B 2000 and LEVURE IOC 18-2007 in races 10.5 % vol. LEVURE IOC B 3000-9.6 % vol. Organic acids affect the taste and bouquet of wine, determine the direction of biochemical processes. As a result of the conducted studies (Table. 1) it was found that the concentration of titrated acids decreased, especially in the case of the use of the yeast LEVURE IOC B 2000.

Table 1. - Mass concentration of organic acids depending on the yeast race

Indicators	The yeast race			
	Control	LEVURE IOC B 3000	LEVURE IOC B 2000	LEVURE IOC BE THIOLS
Mass concentration of organic acids, g/dm <sup>3</sup>				
Wine	2,43	2,20	2,18	2,24
Apple	1,93	1,78	1,85	1,68
Amber	0,32	0,58	0,66	0,45
Lemon	0,32	0,26	0,44	0,18
Acetic acid	0,28	0,25	0,40	0,36
Dairy	no	1,26	0,86	0,47
Tasting score, score	7,6	7,8	7,7	8,1
Mass concentration of titrated acids, g / dm <sup>3</sup>	7,2	6,8	7,0	6,7

The content of tartaric acid varied slightly, while its decrease was noted when using the LEVURE IOC B 2000 and LEVURE IOC B 3000 races. During the fermentation process, a decrease in the concentration of malic acid was revealed. Its amount in wine materials varied from 1.68 g / dm<sup>3</sup> when using the LEVURE IOC BE THIOLS race to 1.93 g/dm<sup>3</sup> in the control. The greatest reduction in the concentration of malic acid was provided by the use of the LEVURE IOC BE THIOLS race.

The results obtained allow us to assume that all experimental yeast races have an acid-reducing ability, which is especially important and significant at the present time, when additional technological operations have to be carried out to reduce the concentration of titrated acids. The use of the studied

yeast races in the technology makes it possible to regulate the concentration of titrated acids already at the stage of alcoholic fermentation. In comparison with the control, the experimental samples showed the accumulation of succinic acid, especially in the variants obtained using the yeast races LEVURE IOC B 2000 and LEVURE IOC B 3000 (0.58 and 0.66 g/dm<sup>3</sup>, respectively). It is known [8] that the presence of succinic acid increases the antioxidant capacity of white table wines and their resistance to oxidation. Consequently, the samples obtained with LEVURE IOC B 2000 and LEVURE IOC B 3000 are more resistant to oxidation in comparison with other wine materials. An increase in lactic acid content was observed in all samples, especially when using the LEVURE IOC B 2000 and LEVURE IOC BE THIOLS strains (0.47 and 0.88 g/dm<sup>3</sup>, respectively). No lactic acid was detected in the control sample. This indicates that the LEVURE IOC 18-2007 yeast race does not contain enzyme systems that transform malic acid to lactic acid [9].

Acetic acid is rapidly formed at the beginning of fermentation, and by the end of fermentation, its content can drop sharply. Thus, studies [10] have shown that this property of yeast is genetic and can increase or decrease depending on the chemical composition of the wort. Studies have shown that the lowest concentration of acetic acid was found when using the LEVURE IOC 18-2007 and LEVURE IOC B 3000 races. Based on the research of Saenko N.F. [11], it is possible to recommend these yeast races for the treatment of "sick" wines with an increased concentration of acetic acid and acetaldehyde.

Oxalic acid was also detected in three wine materials, which indicates different activity of the Krebs cycle enzyme systems. This fact is confirmed by a large difference in the accumulation of pyruvic acid, through which most of the transformations of organic acids in the tricarboxylic acid cycle occur [16].

Citric acid is a natural by-product of alcoholic fermentation [15] and is involved in the addition of wine taste, redox processes. It is formed in the Krebs cycle from pyruvic acid under the action of acetyl-CoA or from oxalic-acetic acid under the action of decarboxylating enzymes of the Krebs cycle. It was found that the synthesis of citric acid of the studied yeast race can be arranged in a row (in descending order of concentrations) LEVURE IOC B 2000 > LEVURE IOC 18-2007 > LEVURE IOC B 3000 > LEVURE IOC BE THIOLS.

Organic acids are involved in esterification reactions, the product of which are various esters and esters that form the aroma of wine. The activity of esterification is determined by the enzymatic complex of yeast cells. Analysis of research materials (Table. 2) indicates that the most important flavor-forming compounds vary widely depending on the yeast race. It was found that the fermentation of wort by LEVURE IOC B 2000 and LEVURE IOC



B 3000 strains led to an increase in the accumulation of acetaldehyde, which can give excessive sharpness and coarseness to the aroma and taste of the wine.

Table 2. - Accumulation of aroma-forming components, mg/dm<sup>3</sup>, depending on the yeast race

Indicators	The name of the yeast			
	LEVURE IOC 18-2007	LEVURE IOC B 3000	LEVURE IOC B 2000	LEVURE IOC BE THIOLS
Acetaldehyde	80,5	121,3	68,7	81,4
Ethyl Formate	0,21	0,22	no	0,18
Ethyl Acetate	23,45	19,78	44,8	36,5
Isoamyl Acetate	0,46	0,51	0,19	0,32
Ethyl Lactate	0,13	0,20	0,22	0,27
Ethyl Acetal	no	0,04	no	0,01
Methanol	20,0	24,1	18,6	21,9
2-butanol	no	0,12	no	0,08
1-propanol	19,4	20,6	15,8	21,2
Isobutanol	59,7	60,1	32,7	41,0
1-butanol	1,12	1,32	no	no
Isoamyl alcohol	168	171	212	186
1-amyl alcohol	0,36	0,38	no	0,12
1-hexanol	18,6	18,5	16,6	18,3
2-phenylethanol	21,6	38,9	32,5	31,2
2,3-butanediol	38,7	58,0	40,4	39,4
Tasting grade, score	7,8	8,2	8,2	8,4

It is known [3, 6] that acetaldehyde is a secondary product of alcoholic fermentation, and its concentration is largely determined by the amount of sulfur dioxide in the initial wort. In our experiments, the initial wort had the same concentration of sulfur dioxide, which was 70 mg/dm<sup>3</sup>, therefore, the differences in the concentrations of acetaldehyde in the experimental wine materials are caused by other factors, namely, the specifics of the yeast race. In addition, during fermentation, a constant new formation of acetaldehyde is usually observed due to deamination of amino acids [14], the activity of which is determined by the biosynthetic functions of the corresponding yeast enzyme systems.

A significant accumulation of isoamyl acetate and 2,3-butanediol, which have an adverse effect on the taste and aroma of wine, was found in wine materials prepared using LEVURE IOC B 3000 yeast races. A significant difference between the races was observed in the number of terpene compounds - citral, terpeniol, limonene, and ionone. Perhaps it is their presence that can explain the bright floral aroma of wine materials prepared using the LEVURE IOC BE THIOLS and LEVURE IOC B 2000 strains.

As a result of analyses of wine materials

produced using the LEVURE IOC BE THIOLS and LEVURE IOC B 2000 strains, it was found that the concentration of isoamyl alcohol was slightly higher, and ethyl acetal, on the contrary, accumulated a small amount compared to the control sample. According to the concentration of ethyl lactate, the samples of wine materials obtained using the IOC BE THIOLS and LEVURE IOC B 2000 differ favorably from the control.

The most important among the esters is ethyl acetate, which has a light fruity aroma, and ethyl lactate, which softens the taste of table wine material. As a result of the studies, the largest amount of ethyl acetate was found in the wine material obtained by fermentation of the wort with the LEVURE IOC B 2000 yeast race, a slightly smaller accumulation was obtained when using the IOC BE THIOLS race. In the other variants, including the control, the content of ethyl acetate had similar values. The greatest accumulation of ethyl lactate was found in the variants using the IOC BE THIOLS, LEVURE IOC B 2000 and LEVURE IOC B 3000 races.

The mass concentration of isoamyl acetate, which shows floral and bitter-fruit tones depending on the concentration, was the highest when using the IOC BE THIOLS and in the control.

Also presented in table aromaterapia components identified in the wine capric, Caproic aldehyde, ethylbutyrate, ethylmalonic, Edilkamin, ethylcaproic, eticaret, isoamylene, esters of linoleic, linolenic, oleic and myristic acids, isovalerianate, capric, Caproic and Caprylic acids, the concentration of which had similar values for all the studied races of yeast. A comparative analysis of the accumulation of higher alcohols is of interest.

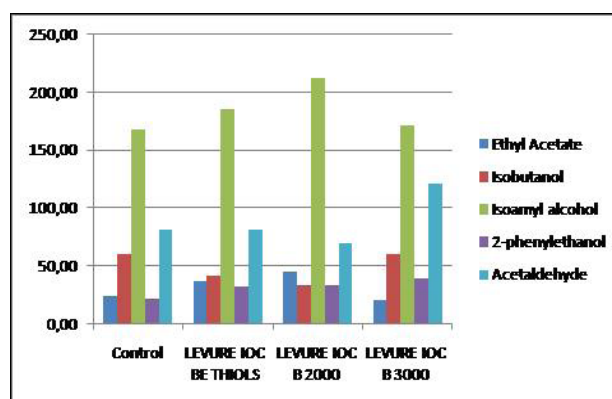


Figure 2. Changes in the concentration of aromatic components depending on the yeast race

They are formed by deamination or transamination of the corresponding amino acids, subsequent decarboxylation of ketoacids, and reduction of aldehydes during alcoholic fermentation [13]. According to [17], higher alcohols are products of alcoholic fermentation, the amount of which is determined by the yeast race, and the synthesis is carried out on the verge of carbohydrate and nitrogen exchange of yeast, i.e., the synthesis of higher

alcohols is determined by the genetic characteristics of yeast.

The analysis of the obtained experimental data indicates that under the same fermentation conditions, 2-butanol was detected only when using the IOC BE THIOLS and LEVURE IOC B 3000 races (Fig. 2). Both the experimental and control yeast races were characterized by a low synthesis of 1-propanol and 1-butanol, which is a positive factor.

Of the known higher alcohols of grape processing products, isoamyl alcohol has the most unpleasant aroma. Its accumulation was greatest when using LEVURE IOC B 2000 yeast races, which, however,

did not have a negative impact on the quality of the wine material.

**Conclusion.** Tasting wine materials was successfully carried out, which revealed a correlation between sensory evaluation and the concentration of titratable acids, especially malic and lactic, and higher alcohols. The wine material prepared using the LEVURE IOC BE THIOLS race had the highest tasting score, while the other wine materials had similar organoleptic scores.

Thus, the presented research materials indicate a significant influence of the yeast race on the aromatic complex of white table wine materials.

**З.А. Анарбекова\*, Г.И. Байгазиева**

Алматы технологиялық университеті, Алматы, Қазақстан.

E-mail: zaure96\_96@mail.ru

### **АШЫТҚЫ НӘСІЛДЕРІНІҢ ҮСТЕЛ ШАРАПТАРЫНЫҢ ХОШ ИІСТІ ЗАТТАРЫНА ӘСЕРІ**

**Аннотация.** Шарап - бұл бақыланатын алкогольдік ашыту, яғни ашыту арқылы жүзім шырынында болатын қосылыстардың, биохимиялық өзгерістердің өнімі. Жүзім мен ашытқы ферменттері жүзімді өңдеуде және шарап дайындауда шешуші рөл атқарады, шарап жасаудың барлық биотехнологиялық процестеріне әсер етеді.

Сусынға сұйық немесе құрғақ белсенді ашытқыны қосу ашыту процесін жақсы басқаруға мүмкіндік береді. Осы ашытқылардың әсерінен қант негізінен алкогольге немесе көмірқышқыл газына айналады, бірақ ашытқы өзі ашыту кезінде шараптың иісі мен дәміне әсер ететін көптеген молекулаларды (жоғары спирттер, эфирлер) түзеді. Бұл түрлендірулер шамамен екі аптаға созылады және температураның едәуір жоғарылауына әкеледі, оны 18-20°C-тан жоғары көтерілуіне жол бермей, оны реттеу керек: әйтпесе хош иісті заттардың кейбіреулері булануы мүмкін және ашыту процесінің өзі тоқтайды.

Дұрыс және толық ашытуды анықтайтын ашытқының мөлшері сусынның сапасына да, ауаның, қоршаған ортаның температурасына да азды-көпті қол жетімділігіне байланысты. Ауа, дәлірек айтқанда, ауаның оттегі ашытқыларға пайдалы әсер етеді, егер сулада әлі де көптеген қоректік заттар (қанттар) болса; соңғысы тұтынылғандықтан, лайлану түрінде ұзақ уақыт сақталатын өте кішкентай ашытқы жасушалары түзіледі. Ашытудың жылдам дамуына прогресті ашытқыны мезгіл-мезгіл араластыру арқылы едәуір жеңілдетуге болады, ол түбіне дейін түсіп, қоректік заттармен тікелей байланысын жоғалтады - төменгі қабаттар дерлік жұмыс істемейді. Сіз сусланы механикалық түрде немесе оған толық жүзім қосу арқылы араластыра аласыз; сонымен бірге сула үнемі және автоматты түрде араласады: жидектер ашыту сұйықтығында көтеріліп, өздерімен бірге ашытқыны алып жүреді. Ашытуды тездету үшін сусланы кейде желдетеді, яғни оған араластыру арқылы ауа енгізеді.

Бұл мақалада ашытқы нәсілінің ақ жүзімнің ашыту динамикасына, органикалық қышқылдардың құрамына және хош иістендіргіш компоненттердің әсері көрсетілген. Жоғары сапалы шарап материалдарын ұсынатын жарыстар ерекше атап өтілді.

**Түйін сөздер:** Шарап ашытқысы, ашыту, ферменттердің белсенділігі, ашыту динамикасы, органикалық қышқылдар, хош иісті заттар.

**З.А. Анарбекова\*, Г.И. Байгазиева**

Алматинский технологический университет, Алматы, Казахстан.

E-mail: zaure96\_96@mail.ru

### **ВЛИЯНИЕ РАС ДРОЖЖЕЙ НА АРОМАТООБРАЗУЮЩИЕ ВЕЩЕСТВА СТОЛОВЫХ ВИН**

**Аннотация.** Вино является продуктом биохимических превращений, присутствующих в виноградном соке путем контролируемой алкогольной ферментации, другими словами брожения. Ферменты винограда и дрожжей играют главную роль в обработке винограда и приготовлении вина и влияют на все биотехнологические процессы в виноделии.

Добавление жидких или сухих активных дрожжей в сусло позволяет лучше держать под контролем

процесс ферментации. Под влиянием этих дрожжей сахар превращается в основном в спирт или углекислый газ, но сами дрожжи во время брожения производят множество молекул (высшие спирты, эфиры), которые влияют на аромат и вкус вина. Эти превращения занимают около двух недель и приводят к существенному увеличению температуры, которую необходимо регулировать, не давая ей подняться выше 18-20°C: в противном случае некоторые ароматические вещества могут улетучиться и сам процесс брожения остановится. Число дрожжей, определяющее правильное и полное брожение, зависит как от качества самого сусла, так и от более или менее продолжительного доступа воздуха, температуры окружающей среды.

Воздух, а точнее кислород в воздухе, благотворно влияет на ферментацию, если в сусле еще много питательных веществ (сахаров); по мере употребления последних образуются очень мелкие дрожжевые клетки, которые сохраняются в течение длительного времени в виде помутнения. Быстрому прогрессу брожения в значительной степени способствует периодическое перемешивание дрожжей, которые, оседая вниз, теряют прямой контакт с питательными веществами - нижние слои почти не функционируют. Перемешивать сусло можно механически или добавлением в него здоровых цельных ягод винограда; при этом сусло постоянно и автоматически перемешивается: ягоды, поднимаясь в бродящей жидкости вверх, увлекают за собой дрожжи. С целью ускорения брожения сусло иногда проветривают, то есть вводят в него воздух, путём перемешивания.

В данной статье показано влияние расы дрожжей на динамику брожения белого виноградного сусла, состав органических кислот и ароматобразующих компонентов. Выделены расы, обеспечивающие получение высококачественных виноматериалов.

**Ключевые слова:** винные дрожжи, брожение, активность ферментов, динамика брожения, органические кислоты, ароматобразующие вещества.

#### Information about authors:

Anarbekova Zauriya – undergraduate, Almaty Technological University, e-mail: zaure96\_96@mail.ru, <https://orcid.org/0000-0002-9137-0634>.

Baigaziyeva Gulgaisha – PhD, associate Professor, Department of Technology of bread products and processing industries, Almaty Technological University, e-mail: bgulgaishailias@mail.ru, <https://orcid.org/0000-0002-9163-4767>.

#### REFERENCES

- [1] Ageeva N.M., Avanesyants R. V. (2017) Influence of enzyme preparations on wine biopolymers / Fruit growing and viticulture in the South of Russia. N 46 (04). P. 129-140. (in Russ).
- [2] Bushina E. V. (2012) New highly effective enzyme preparations for hydrolysis of pectin-and cellulose-containing substrates based on recombinant strains of fungi of the genus *Penicillium*: dis. cand. chem. sci. - Moscow: Lomonosov Moscow State University. P. 114-118. (in Russ).
- [3] Abbes F., Bouaziz M., C. Zoss, Masmoud M., Attia H., Besbes S. (2011) Date syrup: Effect of hydrolytic enzymes (pectinase/cellulase) on the physicochemical characteristics, sensory and functional properties // *Journal of Food Science and Technology*. V. 44. –P. 1827-1834. (in Eng).
- [4] Technochemical control methods in winemaking // Under the editorship of V. G. Gerjikov. 2nd ed. - Simferopol: Tavrida. (2009). p. 304. (in Russ).
- [5] Yadav S., Yadav P. K., Yadav D., Yadav K. D. S. Pectin-lyase: a review // *Process Biochemistry*. (2009). –V. 44. –P. 1-10. (in Eng).
- [6] Jayani R.S., Saxena S., Gupta R. (2003) Microbial pectinolytic enzymes: a review // *Process Biochemistry*. V 40. - P. 2931-2944. (in Eng).
- [7] Kushnereva E.V., Guguchkina T. I., Yakuba Yu. F., Pankin M.I., Lukyanov A. A., Maksimov R. A. (2013) Formation of the quality of table red wines when processing pulp with enzyme preparations. No. 6. – p. 25-29. (in Russ).
- [8] Shatirishvili Sh. I., Makharoblilze M.R., Chkhikvadze Kh. Sh., Tsereteli B.S. (2011) Influence of various technological methods and pasting materials on the colloidal stability of wine // *News of agrarian science*. - No. 3. - p. 94-96. (in Russ).
- [9] Negro G., Hannan M.T., Rao H. (2011) Category reinterpretation and defection: modernism and tradition in Italian winemaking // *Organization Science*. –V. 22. – No. 6. –P. 1449-1463. (in Eng).
- [10] Baffi M. A., Tobal T., Lago, J. H., M. Boscolo, E. Gomes, Da-Silva R. (2013) improvement of Wine aroma using a  $\beta$ -Glucosidase preparation from *Aureobasidium pullulans* // *Appl Biochem Biotechnol*. –V. 169. –P. 493-501. (in Eng).
- [11] Ageeva N. M. (2007) Stabilization of grape wine: theoretical aspects and practical recommendations. - Krasnodar: Prosveshchenie-Yug, - 251 p.
- [12] Puertas B., Jimenez – Hierro M.J. (2018) The influence of yeast on chemical composition and sensory properties of dry white wines // *Food Hydrocolloids*, Vol. 253. <https://doi.org/10.1016/j.foodchem.2018.01.039>

- [13] Moreno-Arriba M., Polo, M. (2005) Winemaking Biochemistry and Microbiology: Current Knowledge and Future Trends // *Critical Reviews in Food Science and Nutrition*. No. 4. - PP. 265-286.
- [14] Konrad V.Miller., David E.Block. (2020) A review of wine fermentation process modeling // *Journal of Food Engineering*, Vol. 273. <https://doi.org/10.1016/j.jfoodeng.2019.109783>
- [15] D'agostino S., E. Carruba, V. Pastena, Alagna C. gascromatografico Studio di alcunicomponenti secondari di vini Siciliani // *Riv. Viticolt. E Ind.* (2009). Vol 32, №12.-P. 496-513. 153. (in Eng).
- [16] Eze M.O., OganUr.A. (2008) Sugar of the in fermented sap and wine from the oil palm *Klaucisguinensis*, tree // *Plant foods Human Nutr* Vol.38.- №2.-P. 121- 126. 154. (in Eng).
- [17] Filajdic M., Djukovic J. (2013) Gas-chromatographic determination of volatile constituents in Yugoslav plum brandies // *J. Sci. Food and agr.* – Vol 24, No. 7. - P. 835-842. 155. (in Eng).

## NEWS

OF THE NATIONAL ACADEMY OF SCIENCES OF THE REPUBLIC OF KAZAKHSTAN  
SERIES CHEMISTRY AND TECHNOLOGY

ISSN 2224-5286

Volume 3, Number 447 (2021), 18 – 24

<https://doi.org/10.32014/2021.2518-1491.44>

UDC 542.943; 547.211; 547.313.2  
IRSTI 61.51.21

**D.B. Augaliev<sup>1</sup>, M.K. Erkibaeva<sup>2</sup>, A.O. Aidarova<sup>3</sup>,  
S.A. Tungatarova<sup>4</sup>, T.S. Baizhumanova<sup>1,4</sup>**

<sup>1</sup>Al-Farabi Kazakh National University, Almaty, Kazakhstan;

<sup>2</sup>Pavlodar State Pedagogical University, Pavlodar, Kazakhstan;

<sup>3</sup>M.H. Dulati Taraz Regional University, Taraz, Kazakhstan;

<sup>4</sup>D.V. Sokolsky Institute of Fuel, Catalyst and Electrochemistry, Almaty, Kazakhstan.

E-mail: [augaliev.diar.97@gmail.com](mailto:augaliev.diar.97@gmail.com)

## OXIDATIVE DIMERIZATION OF METHANE TO C<sub>2</sub> HYDROCARBONS

**Abstract.** The world's oil reserves are decreasing every day due to the continuous production and processing of the most modern technologies. Scientists all over the world are looking for various raw materials and methods to use the vast resources of natural gas as a substitute for petrochemicals. In this regard, great attention is drawn to natural gas as an alternative source of raw materials for petrochemical industries.

The purpose of this work is to study the reaction of methane dehydrogenation on new 20%La-10%Ce-20%Mg-50% glycine catalysts prepared by the SHS method to identify the optimal conditions for their preparation, concentration and ratio of metals, the influence of contact time and process temperature on the direction and mechanism of the reaction.

The results of the study of 20% La-10% Ce-20% Mg-50% glycine catalyst prepared by the SHS method in the process of oxidative dehydrogenation of methane into C<sub>2</sub> hydrocarbons are presented. On the basis of experimental studies, it was found that the composition of the catalyst exhibits high activity in the above reaction under the found optimal conditions.

Thus, the influence of reaction temperature on the developed composition of catalysts for oxidative conversion of methane has been determined that the optimum temperature for the selective formation of ethane and ethylene is T=700°C. It was found that for selective oxidation of a mixture of CH<sub>4</sub>: O<sub>2</sub> : Ar in C<sub>2</sub> hydrocarbons the optimal conditions are: T=700°C, CH<sub>4</sub>:O<sub>2</sub>=2,5:1, 5000 h<sup>-1</sup>.

**Key words:** ODM reaction, C<sub>2</sub> hydrocarbons, SHS method, temperature of reaction.

**Introduction.** The world's oil reserves are decreasing every day due to the continuous production and processing of the most modern technologies. Scientists all over the world are looking for various raw materials and methods to use the vast resources of natural gas as a substitute for petrochemicals. In this regard, great attention is drawn to natural gas as an alternative source of raw materials for petrochemical industries.

According to the annual Statistical Review of

World Energy – 2018, published by British Petroleum (BP) at the end of 2017 proven natural gas reserves in the world are estimated at 193.5 trillion cubic meters.

It is known that natural gas is 90% methane. Of interest is the oxidative dimerization of methane into ethylene, which makes it possible to obtain a number of petrochemical products, such as polyethylene, polystyrene and many other chemical products. This is primarily due to the low cost of methane compared to other hydrocarbons [1].

---

**МАЗМУНЫ – СОДЕРЖАНИЕ – CONTENTS**


---

Аппазов Н.О., Диярова Б.М., Базарбаев Б.М., Асылбекқызы Т., Джиембаев Б.Ж. КҮРШ ҚАЛДЫҒЫМЕН МҰНАЙ ШЛАМЫН БІРГЕ ӨНДЕУДЕ БАЙЛАНЫСТЫРУШЫ КРАХМАЛ НЕГІЗІНДЕ БРИКЕТТЕЛГЕН БЕЛСЕНДІРІЛГЕН КӨМІР АЛУ.....	6
Anarbekova Z.A., Baigazieva G.I. THE INFLUENCE OF YEAST RACES ON THE AROMA-FORMING SUBSTANCES OF TABLE WINES.....	12
Augaliev D.B., Erkibaeva M.K., Aidarova A.O., Tungatarova S.A., Baizhumanova T.S. OXIDATIVE DIMERIZATION OF METHANE TO C2 HYDROCARBONS.....	18
Әбдібек А.Ә., Мулдабекова Б.Ж., Якияева М.А., Идаятова М.А., Әбіл А.Ж. ҰНДЫ КОНДИТЕР ӨНДІРІСІНДЕ ДӘСТҮРЛІ ЕМЕС ШИКІЗАТТАРДЫ ҚОЛДАНУДЫҢ ТИІМДІЛІГІ.....	24
Исаева Н.А., Байгазиева Г.И. ҚАНТ АЛМАСТЫРҒЫШ – СТЕВИЯ [STEVIA REBAUDIANA BERTONI (L.)] ҚОСЫЛҒАН СУЫҚ ҚАРА ШАЙ ӨНДІРІСІ.....	31
Ikhsanov Y.S., Kusainova K.M., Tasmagambetova G.Y., Andasova N.T., Litvinenko Y.A. AMINO ACID, FATTY ACID AND VITAMIN COMPOSITION OF ROSA CANINA L.....	39
Jalmakhanbetova R.I., Suleimen Ye.M., Kasenov B.K. CALCULATE THE STANDARD ENTHALPIES OF COMBUSTION, FORMATION AND MELTING OF THE COMPLEX ROSEOFUNGIN WITH $\alpha$ -, $\beta$ - and $\gamma$ -CYCLODEXTRIN.....	44
Kairbekov Zh.K., Jeldybayeva I.M., Abilmazhinova D.Z., Suimbayeva S.M. PHYSICOCHEMICAL AND ANTIOXIDANT PROPERTIES OF HUMIC ACIDS OF LOW-MINERALIZED PELOIDS OF THE TUZKOL DEPOSIT.....	48
Mamyrbekova Aizhan, Mamyrbekova Aigul, Kassymova M.K., Aitbayeva A.Zh., Chechina O.N. STUDY OF KINETICS OF COPPER OXIDATION BY ELECTROLYSIS UND NON-STATIONARY CONDITIONS.....	54
Madet G., Bayazitova M.M. RESEARCH OF MALTING PROPERTIES OF KAZAKHSTAN TRITIKALE GRAIN VARIETIES FOR USE IN THE BEVERAGE INDUSTRY.....	59
Mussina A.S., Baitasheva G.U., Myrzakhmetova N.O., Tagabergenova Zh.A., Gorbulicheva E.P. EVELOPMENT OF THE CONDITIONS FOR STORING THALLIUM AMALGAM.....	65
Naguman P.N., Zhorabek A.A., Amanzholova A.S., Kulakov I.V., Rakhimbaeva A.N. PHYTONCIDES IN THE COMPOSITION OF COMMON BIRD CHERRY.....	70
Nurdillayeva R.N., Sauribay Zh.G., Bayeshov A.B. DISSOLUTION OF STAINLESS STEEL IN SODIUM CHLORIDE SOLUTION AT POLARIZATION BY NON-STATIONARY CURRENT.....	75
Rakhimberlinova Zh.B., Kulakov I.V., Alimzhanova A.Zh., Mussirepov M.M., Nakypbekova N.E. CHEMICAL ACTIVATION OF THE SURFACE OF THE BURNT ROCK BY VARIOUS MODIFIERS.....	81
Rasulov S.R., Mustafayeva G.R. DEVELOPMENT OF EFFECTIVE CATALYSTS FOR PROCESSING C3-C4 HYDROCARBONS.....	87

Tyan A., Bayazitova M.M. SELECTION OF THE MASHING MODE IN THE PREPARATION OF BEER WORT BY USING THE WHEAT MALT.....	94
Vysotskaya N.A., Kabyzbekova B. N., Spabekova R.S., Asylbekova D.D., Lukin E.G. SOME FEATURES OF ELECTRODEPOSITION OF METALS FROM ELECTROLYTES WITH SURFACTANTS.....	99
Yermagambet B.T., Kazankapova M.K., Nauryzbayeva A.T., Kassenova Zh.M. SYNTHESIS OF CARBON NANOFIBERS BASED ON HUMIC ACID AND POLYACRYLONITRILE BY ELECTROSPINNING METHOD.....	103

---

**Publication Ethics and Publication Malpractice  
in the journals of the National Academy of Sciences of the Republic of Kazakhstan**

---

For information on Ethics in publishing and Ethical guidelines for journal publication see <http://www.elsevier.com/publishingethics> and <http://www.elsevier.com/journal-authors/ethics>. Submission of an article to the National Academy of Sciences of the Republic of Kazakhstan implies that the described work has not been published previously (except in the form of an abstract or as part of a published lecture or academic thesis or as an electronic preprint, see <http://www.elsevier.com/postingpolicy>), that it is not under consideration for publication elsewhere, that its publication is approved by all authors and tacitly or explicitly by the responsible authorities where the work was carried out, and that, if accepted, it will not be published elsewhere in the same form, in English or in any other language, including electronically without the written consent of the copyright holder. In particular, translations into English of papers already published in another language are not accepted.

No other forms of scientific misconduct are allowed, such as plagiarism, falsification, fraudulent data, incorrect interpretation of other works, incorrect citations, etc. The National Academy of Sciences of the Republic of Kazakhstan follows the Code of Conduct of the Committee on Publication Ethics (COPE), and follows the COPE Flowcharts for Resolving Cases of Suspected Misconduct ([http://publicationethics.org/files/u2/New\\_Code.pdf](http://publicationethics.org/files/u2/New_Code.pdf)). To verify originality, your article may be checked by the Cross Check originality detection service <http://www.elsevier.com/editors/plagdetect>. The authors are obliged to participate in peer review process and be ready to provide corrections, clarifications, retractions and apologies when needed. All authors of a paper should have significantly contributed to the research.

The reviewers should provide objective judgments and should point out relevant published works which are not yet cited. Reviewed articles should be treated confidentially. The reviewers will be chosen in such a way that there is no conflict of interests with respect to the research, the authors and/or the research funders.

The editors have complete responsibility and authority to reject or accept a paper, and they will only accept a paper when reasonably certain. They will preserve anonymity of reviewers and promote publication of corrections, clarifications, retractions and apologies when needed. The acceptance of a paper automatically implies the copyright transfer to the National Academy of Sciences of the Republic of Kazakhstan.

The Editorial Board of the National Academy of Sciences of the Republic of Kazakhstan will monitor and safeguard publishing ethics.

Правила оформления статьи для публикации  
в журнале смотреть на сайте:

**[www:nauka-nanrk.kz](http://www.nauka-nanrk.kz)**

**<http://chemistry-technology.kz/index.php/en/arhiv>**

**ISSN 2518-1491 (Online), ISSN 2224-5286 (Print)**

Редакторы: *М.С. Ахметова, Р.Ж. Мрзабаева, Д.С. Аленов*  
Верстка на компьютере *В.С. Зикирбаева*

Подписано в печать 12.06. 2021.  
Формат 60x881/8. Бумага офсетная. Печать – ризограф.  
10,2 п.л. Тираж 300. Заказ 3.

---

*Национальная академия наук РК*  
*050010, Алматы, ул. Шевченко, 28, т. 272-13-18, 272-13-19*